Why Formic Acid?

Bioenergy Bandung Insitute of Technology (Indonesia) Chemurgy

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Clear liquid, pungent odor, sharp taste;
corosive (rather strong acid);

• $T_{\text{freezing}} = 8,4 \text{ °C};$ • $T_{\text{boiling}} = 101 \text{ °C};$ • $\rho = 1,22 \text{ kg/L};$

Molecular weight M_r = 46,03 g/mol.
2 gram of H₂ in 38 ml liquid!.



Routes of formic acid production





25 kW formic acid fuel cell electricity generating unit

Formic acid fuel cell powered bus

Hydrogen generation from formic acid (in the vehicles fuel tank)

• In contact with noble metal catalysts (Pd > Pt > Au > Ag) nearby room temperature, formic acid will decomposes into hydrogen and carbon dioxida (CO₂):

 $HCOOH_{(liquid)} \longrightarrow H_{2(gas)} + CO_{2(gas)}$

- If necessary, the hydrogen produced can be separated from its mixture with CO_2 using membrane technology.
- The (purified) hydrogen can then be burned in the IC engine.

Bioenergies Engineering and Chemurgy

Review

Hydrogen Internal Combustion Engine Vehicles: A Review

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THANK YOU VERY MUCH for your attention Bioenergy Engineering and Chemurgy